

inférieure à 90° lorsque les composés possèdent les groupes spatiaux $R3m$ ou $R3c$ (Megaw & Darlington, 1975).

La valeur de la distance Al—O trouvée égale à 1,892 Å, permet de calculer le paramètre a d'une maille cubique perovskite hypothétique: $a = 2d(\text{Al—O}) = 3,784$ Å. On en déduit dans ce cas la distance Nd—O = $a/\sqrt{2} = 2,676$ Å dans un cubo-octaèdre NdO_{12} régulier ainsi que les douze distances O—O dans l'octaèdre AlO_6 , égales aussi à 2,676 Å. La déformation de la structure conduit à observer deux sortes de distances O—O. La première, égale à 2,696 Å, concerne les atomes situés dans des plans perpendiculaires à l'axe [111]; la seconde, égale à 2,655 Å et relative à des distances entre ces plans, est plus courte et caractérise l'aplatissement de l'octaèdre.

Le cubo-octaèdre NdO_{12} est également déformé et on observe trois types de liaisons Nd—O. Perpendiculairement à l'axe [111], il existe six oxygène qui forment avec le néodyme un plan. De plus, trois oxygène sont situés dans un plan supérieur et trois autres dans un plan inférieur. La distance entre Nd et ces six derniers oxygène est égale à 2,656 Å. Dans le plan médian, on observe trois distances Nd—O plus courtes = 2,406 Å et trois oxygène significativement plus éloignés = 2,914 Å. Il en résulte que la coordinence du néodyme tend vers neuf, valeur qu'on retrouve dans le composé $\text{Nd}_2\text{AlO}_3\text{N}$

(Marchand, Pastuszak, Laurent & Roul, 1982) qui possède une structure de type K_2NiF_4 , donc apparentée à la structure perovskite.

Références

- BOUCHERLE, J. X. & SCHWEIZER, J. (1975). *Acta Cryst.* **B31**, 2745–2746.
 BURBANK, R. D. (1970). *J. Appl. Cryst.* **3**, 112–120.
 DERIGHETTI, B., DRUMHELLER, J. E., LAVES, F., MÜLLER, K. A. & WALDNER, F. (1965). *Acta Cryst.* **18**, 557.
 GELLER, S. & BALA, V. B. (1956). *Acta Cryst.* **9**, 1019–1025.
 JACOBSON, A. J., TOFIELD, B. C. & FENDER, B. E. F. (1972). *Acta Cryst.* **B28**, 956–961.
 KHATTAK, C. P. & WANG, F. F. Y. (1979). *Handbook on the Physics and Chemistry of Rare Earths*, Tome 3, édité par K. A. Gschneidner & L. Eyring, p. 525. Amsterdam: North-Holland.
 KJEMS, J. K., SHIRANE, G., MÜLLER, K. A. & SCHEEL, H. J. (1973). *Phys. Rev. B*, **8**, 1119–1124.
 MARCHAND, R., PASTUSZAK, R., LAURENT, Y. & ROULT, G. (1982). *Rev. Chim. Minér.* **19**, 684–689.
 MEGAW, H. D. & DARLINGTON, C. N. W. (1975). *Acta Cryst.* **A31**, 161–173.
 MENYUK, N., DWIGHT, K. & RACCAH, P. M. (1967). *J. Phys. Chem. Solids*, **28**, 549–556.
 MÜLLER, K. A., BERLINGER, W. & WALDNER, F. (1968). *Phys. Rev. Lett.* **21**, 814–817.
 ROULT, G. & BUEVOZ, J. L. (1977). *Rev. Phys. Appl.* **12**, 581–590.
 WORLTON, T. G., JORGENSEN, J. D., BEYERLEIN, R. A. & DECKER, D. L. (1976). *Nucl. Instrum. Methods*, **137**, 331–337.

Acta Cryst. (1983). **C39**, 675–677

Structures of Five Large-Period Polytypes of Cadmium Iodide

By S. K. CHAUDHARY, G. K. CHADHA AND G. C. TRIGUNAYAT

Department of Physics and Astrophysics, University of Delhi, Delhi-110007, India

(Received 30 September 1982; accepted 7 February 1983)

Abstract. The crystal structures of the five new large-period polytypes $16H_{10}$, $18H_7$, $24H_{10}$, $30H_6$ and $32H_3$ of cadmium iodide have been determined. The structures are represented as $2112(11)_5$, $222112(11)_4$, $(22)_3(11)_31221$, $22(11)_522(11)_6$ and $(22)_6112211$ in Zhdanov notation. They all belong to the space group $P3m1$.

Introduction. The phenomenon of polytypism exhibited by a large number of substances has attracted solid-state physicists because different polytypic modifications of a substance possess different semiconducting, dielectric, optical and photovoltaic properties, which depend upon the crystal structures of the various modifications. With the determination of the complete crystal structures for a large number of polytypes of

prominent polytypic compounds like silicon carbide and cadmium iodide, a vast amount of precise structural data has accumulated, which has already found use in several ways in understanding the formation of polytypes and the phase transitions observed in them. From the analysis of the known crystal structures, the existence of structural series was pointed out by Mitchell (1955, 1956, 1957) in the polytypes of SiC and CdI_2 , viz the series $[(33)_n32]_3$ and $[(22)_n11]$, respectively, the implications of which were discussed later by Trigunayat (1971). The structural data available on the polytypes of SiC, ZnS and CdI_2 have been employed by Srinivasan & Parthasarathi (1973) for a statistical analysis of Zhdanov sequences, to yield semi-empirical, semi-theoretical rules regarding the occurrence and distribution of polytypes. Similarly,

based on the examination of similar data on CdI_2 polytypes, Jain & Trigunayat (1977*a,b*) and recently Wahab & Trigunayat (1980*b*) have formulated empirical guidelines, which can help to reduce the number of probable structures for a given polytype drastically and thus considerably facilitate future crystal structure determinations. From the crystal structure data on the polytypes of SiC, ZnS and CdI_2 , Srinivasan & Parthasarathi (1974) have evaluated the relative proportions of cubic and hexagonal packing in these polytypes and have concluded that they are governed by changes in internal energy, arising from interaction at the atomic level. Similarly, making use of the vast amount of structural data available on the same compounds, Wahab & Trigunayat (1980*a*) have quantitatively estimated the stacking-fault energies and entropy contributions and hence the specific surface free energies of the faults.

So far nearly 289 polytypes of CdI_2 have been reported, of which complete crystal structures of more than 132 polytypes have been worked out (Wahab & Trigunayat, 1980*b*; Chadha, 1982; Palosz, 1982, 1983). In the course of a recent comprehensive study of the influence of impurities on the growth of CdI_2 polytypes, we have discovered several new polytypes. It has been possible to work out complete crystal structures of five of them. The c dimensions of these polytypes are relatively large, ranging from 55.68 to 109.38 Å.

Experimental. The crystals were grown by evaporation of an aqueous solution of reagent-grade cadmium iodide. Details of growth procedure, the selection of crystals and X-ray methods employed are available elsewhere (e.g. Chadha & Trigunayat, 1967). The method of structure determination, together with various refinements, has been described earlier by Jain & Trigunayat (1978), Wahab & Trigunayat (1980*b*) and Chadha (1980).

Discussion. Complete crystal structures of five new CdI_2 polytypes, all hexagonal, have been determined. The X-ray oscillation photographs of the polytypes are reproduced in Fig. 1 and the calculated and observed intensity values of their $10.l$ reflections are listed in Tables 1 to 5.* To ensure uniformity of comparison in these tables, the calculated intensity values have been so adjusted that the reflection $l = 5n/2$ has the value 1000 units for each polytype, where n is the number of layers in the unit cell. The crystal structures, finally arrived at on the basis of obtaining a satisfactory agreement between calculated and observed intensities for $10.l$ reflections, are given in Table 6.

* Tables 1–5 have been deposited with the British Library Lending Division as Supplementary Publication No. SUP 38365 (6 pp.). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

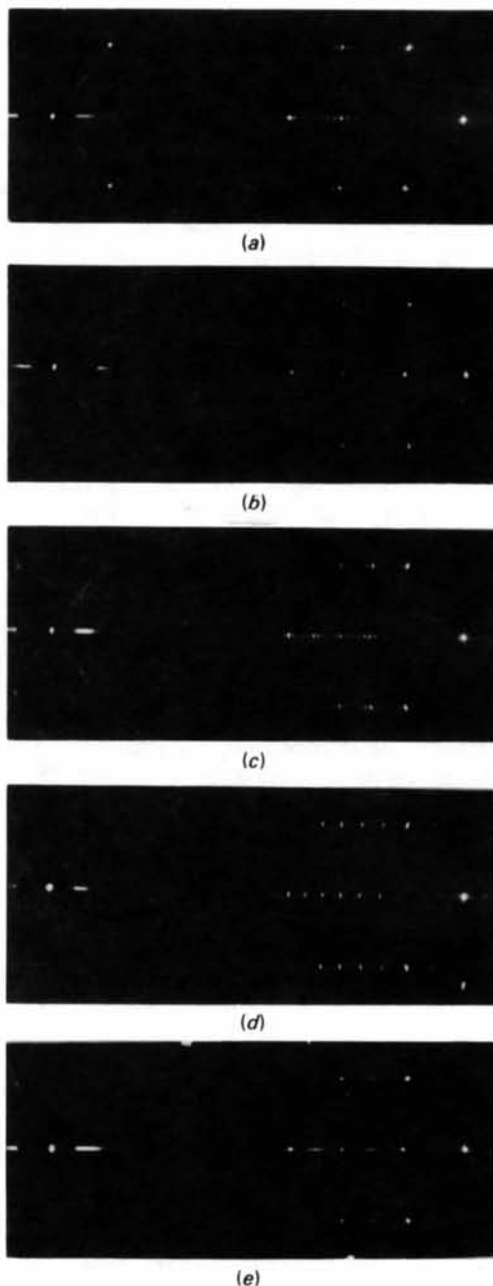


Fig. 1. 15° a -axis oscillation photographs of the polytypes (a) $16H_{10}$, (b) $18H_7$, (c) $24H_{10}$, (d) $30H_6$, (e) $32H_3$; 3 cm camera; Cu $K\alpha$ radiation. The first most intense spot from the centre on the zero layer has $l = 5n/2$ for all five nH polytypes.

Table 6. Detailed crystal structures of new polytypes (space group $P3m1$ and $a=b=4.24$ Å, for all polytypes)

No.	Poly-type	Zhdanov symbol	ABC sequence	c (Å)
1	$16H_{10}$	2112(11) ₅	$A\gamma B (CaB)_2 (A\gamma B)_3$	55.68
2	$18H_7$	222112(11) ₄	$(A\gamma B CaB)_2 CaB (A\gamma B)_4$	61.52
3	$24H_{10}$	(22) ₃ (11) ₁ 1221	$(A\gamma B CaB)_3 (A\gamma B)_4 ABC A\gamma B$	82.02
4	$30H_6$	22(11) ₂ 22(11) ₆	$A\gamma B CaB (A\gamma B)_6 CaB (A\gamma B)_6$	102.54
5	$32H_3$	(22) ₄ 112211	$(A\gamma B CaB)_4 (A\gamma B)_2 CaB A\gamma B$	109.38

Polytype 16H₁₀. This new polytype existed on one basal pinacoidal face of a well developed hexagonal crystal. Though a few-hundred probable structures exist for a 16-layered CdI₂ polytype, the number of possibilities was drastically cut by the observation that the intense reflections lie on or around 2H(11) positions (Fig. 1a), thus suggesting that the Zhdanov sequence of the polytype necessarily contains many (11) units. Further, the intensity distribution closely resembled that of 10H₄, worked out earlier by Jain & Trigunayat (1975). Therefore, the following structures were postulated:

- (1) 22(11)₆
- (2) (22)₂(11)₄
- (3) 2112(11)₅
- (4) 2(11)₂2(11)₄

A satisfactory agreement between the calculated and the observed intensities was found for structure (3) (see Table 1).

Polytype 18H₇. The crystal structures of six 18H polytypes have been reported earlier by various workers. This new modification existed on one of the basal faces of the crystal. The other pinacoid was found to be 4H with some streaking. The *a*-axis oscillation photograph of the polytype is shown in Fig. 1(b). The strong reflections lie on or around 4H positions and the distribution of the reflections is not symmetrical. A large number of stacking sequences are possible for an 18-layered polytype. Intensities were computed for several such sequences, consisting of (22) and (11) units. Of these, the sequence 222112(11)₄ gave a satisfactory agreement with the observed intensities (Table 2).

Polytype 24H₁₀. This structure was found on one of the pinacoids, while the other one showed the reflections of the common polytype 4H. It was observed that the strong reflections either coincided with the 4H positions or lie near them, thus indicating that the unit cell of the structure mainly consists of (22) units. Consequently, a large number of probable sequences containing (22) units were tried. The best matching between the calculated and the observed intensities was found for the structure (22)₃(11)₃1221 (Table 3).

Polytype 30H₆. This polytype existed in coalescence with a much smaller polytype 6H, the crystal structure of which is already known, *viz* (2211). Thus it was suspected that its crystal structure would be based on 6H. A large number of sequences of Zhdanov symbols containing (2211) units were tried, out of which a good agreement between *I*_{calc} and *I*_{obs} for 10.*l* reflections was found for the structure 22(11)₅22(11)₆ (Table 4). It was not possible to compare *I*_{calc} and *I*_{obs} for reflections with *l* = 75, 80, 85, 90, 100 and 105, because of their superposition with the reflections of polytype 6H. However, this was not a serious handicap, since because of the large unit-cell size of the polytype

the total number of reflections available for intensity comparison is still large.

Polytype 32H₃. So far complete crystal structure determinations for only two 32-layered polytypes of cadmium iodide have been reported (Prasad & Srivastava, 1970; Pałosz, 1982). Fig. 1(e) shows the *a*-axis oscillation photograph of the present 32-layered polytype, which was present on one of the pinacoidal faces of a CdI₂ crystal. The other face was identified as 4H. It was observed that strong reflections of 32H either coincide with 4H positions or lie around them, thus indicating that the unit cell consists mostly of (22) units. The following possibilities were tried:

- (1) (22)₇(11)₂
- (2) (22)₆(11)₄
- (3) (22)₆121121
- (4) (22)₆112211
- (5) (22)₅(11)₆
- (6) (22)₅11(22)₂11
- (7) (22)₅12211221
- (8) (22)₅(11)₂22(11)₂
- (9) (22)₅1122(11)₃
- (10) (22)₅112112(11)₂
- (11) (22)₅112(11)₂211
- (12) (22)₅12(11)₃21
- (13) (22)₅1112112111.

Excellent agreement between the observed and calculated intensities was obtained for structure (4) (Table 5).

References

- CHADHA, G. K. (1980). *Acta Cryst.* A36, 1041–1043.
 CHADHA, G. K. (1982). *Acta Cryst.* B38, 3009–3011.
 CHADHA, G. K. & TRIGUNAYAT, G. C. (1967). *Acta Cryst.* 22, 573–579.
 JAIN, P. C. & TRIGUNAYAT, G. C. (1975). *Z. Kristallogr.* 142, 121–126.
 JAIN, P. C. & TRIGUNAYAT, G. C. (1977a). *Acta Cryst.* A33, 255–256.
 JAIN, P. C. & TRIGUNAYAT, G. C. (1977b). *Acta Cryst.* A33, 257–260.
 JAIN, P. C. & TRIGUNAYAT, G. C. (1978). *Acta Cryst.* B34, 2677–2684.
 MITCHELL, R. S. (1955). *Philos. Mag.* 46, 1141–1146.
 MITCHELL, R. S. (1956). *Z. Kristallogr.* 108, 296–315.
 MITCHELL, R. S. (1957). *Z. Kristallogr.* 108, 341–358.
 PAŁOSZ, B. (1982). *Acta Cryst.* B38, 3001–3009.
 PAŁOSZ, B. (1983). *Acta Cryst.* C39, 521–528.
 PRASAD, R. & SRIVASTAVA, O. N. (1970). *Z. Kristallogr.* 131, 376–384.
 SRINIVASAN, R. & PARTHASARTHI, V. (1973). *Z. Kristallogr.* 137, 296–310.
 SRINIVASAN, R. & PARTHASARTHI, V. (1974). *Z. Kristallogr.* 139, 379–394.
 TRIGUNAYAT, G. C. (1971). *Phys. Status Solidi A*, 4, 281–303.
 WAHAB, M. A. & TRIGUNAYAT, G. C. (1980a). *Indian J. Pure Appl. Phys.* 18, 577–582.
 WAHAB, M. A. & TRIGUNAYAT, G. C. (1980b). *Acta Cryst.* A36, 1013–1016.